Biogeochemical modelling with PISCES

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The Pisces model

- Based on the Aumont et al. (2015) paper (no updated documentation yet)
- Coupled with NEMO ocean model and ROMS-AGRIF/CROCO model
- Not all the model is detailed, only the «most important » processes (subjective choice)
- Different versions of PISCES are available depending on the CROCO version :

PISCESv0 in old ROMS-agrif/earlier CROCO versions

PISCESv2 (in newest CROCO version)

=> detailed description of the common parameterizations of PISCESv0 and PISCESv2

+ brief overview of the new potentialities of PISCESv2

=> there may be small differences with parameterizations in Aumont et al. (2015) but the philosophy is the same



The Pisces model

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PISCES-v2: an ocean biogeochemical model for carbon and ecosystem studies

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Pisces model structure

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Equation for Nanophytoplankton (small phytoplankton)



Equation for Nanophytoplankton (small phytoplankton)



Calculation of PAR (Photosynthetically Available Radiation) to compute photosynthesis



Photosynthesis : transformation of mineral/inorganic matter

into living organic matter

Optical model required to compute the penetration of light in the water column

 \Rightarrow 3 wavelengths in PISCES optical model: red, green, blue

 \Rightarrow PAR= PAR_r +PAR_g + PAR_b, Qsol = solar radiation at the surface of the ocean

	red	
Light	$PAR_{r}(z) = PAR_{r}(z - dz) e^{-k_{r}dz}$	
absorption	$PAR_{r}\left(0\right) = \frac{0.43}{3}Q_{sol}$	
Absorption coefficient	$k_r = k_{r0} + \chi_{rp} Pig^{e_r} -$	$\blacktriangleright Pig = \frac{Chl}{r_{pig}}$

Qsol

k_i(z)

PAR

Ζ

Equation for Nanophytoplankton (small phyto cells)

$$\frac{\partial P}{\partial t} = \underbrace{(1 - \delta^{nano})\mu^{nano}P}_{P} - m^{nano}\frac{P}{K_{nano} + P}P - w_p^{nano}P^2 - g^{micro}(P)Z - g^{micro}(P)M$$
Production
$$\mu^{nano} = \mu_P \left(1 - e^{\frac{\alpha^P(\underline{Chl}}{C})P}\underline{P_{AR}}\right) L_{lim}^{nano}$$

$$\mu_P = ab^{cT} \text{ =temperature dependent}$$



Equation for Nanophytoplankton (small phyto cells)

$$\frac{\partial P}{\partial t} = \underbrace{\left(1 - \delta^{nano}\right)\mu^{nano}P}_{\text{Production}} - m^{nano} \frac{P}{K_{nano} + P}P - w_p^{nano}P^2 - g^{micro}(P)Z - g^{micro}(P)M$$

$$\frac{P^{\text{Production}}}{\mu^{nano}} = \frac{\mu_P \left(1 - e^{\frac{\alpha^P \left(\frac{Ohl}{C}\right)P_{PAR}}{\mu_P L_{lim}^{nano}}}\right)}_{\mu_P = ab^{cT} \text{ = temperature dependent}}$$

$$\frac{P^{\text{O4}}}{\mu_P = ab^{cT}} = \underbrace{P^{\text{O4}}}_{Fe} + Fe}_{\text{NO3}} + \underbrace{Fe}_{nano} + Fe}_{NO3} + \underbrace{K_{nan}^{nano}K_{nh4}^{nano} + K_{nh4}^{nano}NO_3 + K_{no3}^{nano}NH_4}}_{\text{NH4}} + \underbrace{L_{nh4}^{nano}}_{Rh4} + K_{nh4}^{nano}NO_3 + K_{no3}^{nano}NH_4} + \underbrace{L_{nh4}^{nano}}_{Rh4} + \underbrace{L_{nh4}^{nano}}_{Rh4} + \underbrace{L_{nano}^{nano}}_{Rh4} + \underbrace{L_{nano}^{nano}$$

Equation for Diatoms (big phyto cells)

$$\frac{\partial D}{\partial t} = \underbrace{(1 - \delta^{diat})\mu^{diat}D}_{\text{Production}} - m^{diat}\frac{D}{K_{diat} + D}D - w_p^{diat}D^2 - g^{micro}(D)Z - g^{meso}(D)M$$

Diatoms need Silicate for their exoskeleton

Limiting nutrients : PO4, NO3, NH4, Fe, Silicate (Si)





Equation for Diatoms (big phyto cells)

$$\frac{\partial D}{\partial t} = (1 - \delta^{diat})\mu^{diat}D - \boxed{m^{diat}\frac{D}{K_{diat} + D}D - w_p^{diat}D^2} - g^{micro}(D)Z - g^{meso}(D)M$$
Mortality
$$\frac{\nabla PO_i^{\circ}}{\nabla PO_i^{\circ}} + \frac{\nabla PO_i^{$$

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Equation for Diatoms (big phyto cells)

$$\frac{\partial D}{\partial t} = (1 - \delta^{diat})\mu^{diat}D - \boxed{m^{diat}\frac{D}{K_{diat} + D}D - w_p^{diat}D^2} - g^{micro}(D)Z - g^{meso}(D)M$$
Two terms : *«* linear *»* and quadratic mortality (= aggregation of cells)

Two terms : « linear » and quadratic mortality (= aggregation of cells)

Aggregation increases when nutrient stress increases, cells become more sticky, and merge into big sinking particles:

$$w_p^{diat} = w_p^{min} + w_p^{max} \times (1 - L_{lim}^{diat})$$



Equation for micro-zooplankton (small cells)



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Equation for micro-zooplankton (small cells)



- p_N=preferential grazing of zoo for species (N) over all species I (= P,D,POCs):

$$g^{micro}(N) = g^{micro} \frac{p_N^{micro} N}{K_G^{micro} + \sum_I (p_I^{micro} I)}$$

 $p_{POC} < p_{Dia} < p_{Nano} = > zoo prefers to graze nanophyto (P), then big phyto (Diatoms), then POC$

- grazing coefficient increases with temperature: same dependence as phytoplankton : b^{cT}



Equation for meso-zooplankton (big cells)



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Equation for meso-zooplankton (big cells)



- Meso zooplankton grazes on two phyto and two detritus size classes (POCs and POCb)

- p^{meso}_N=preferential grazing of M for species (N) over all species I (P,D,POCs,POCb):

$$g^{meso}(N) = g^{meso} \frac{p_N^{meso}N}{K_G^{meso} + \sum_I (p_I^{meso}I)}$$

$$p_N^{meso} = \frac{\gamma_N N}{\sum_I (\gamma_I I)} = \text{not a constant (as for micro zoo)}$$

$$= \text{Meso zoo easts preferentially the most abundant previous of th$$

Equation for dissolved organic matter (DOM) (carbon only : DOM =DOC)



(-Denit if $O_2 low$)



Equation for dissolved organic matter (DOM) (carbon only : DOM =DOC)



Aggregation terms: effect of turbulence=> sh= 1/(time scale): 1s in mixed layer, 100s below mixed layer $\Phi_{\rm e}$ = parameters for probability of encounter of particules

Equation for (small) particulate organic matter (POCs)





Equation for (small) particulate organic matter (POCs)

$$\begin{array}{l} \displaystyle \frac{\partial POC_{s}}{\partial t} &= & \sigma^{micro}(\sum_{N}g^{micro}(N))Z - g^{micro}(POC_{s})Z \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \left(m^{nano}\frac{P}{K_{nano} + P}P + w_{P}^{nano}P^{2}\right) \\ &+ & \left(1 - 0.5R_{CaCO3}\right) \\ &+ & \left(1 - 0.5R_{CCC}\right) \\ &+ &$$

 $w^{POC} = w^{POC}_{min} + (w^{POC}_{max} - w^{POC}_{min}) \max(0, \frac{z - z_{mel}}{2000m}) \qquad \text{Zmel=max(Zmxl,Ze)}$

WPOC

Differences between Nitrogen and Phosphate pools (1)

$$\frac{(\frac{\partial NO_3}{\partial t} + \frac{\partial NH_4}{\partial t}) - \frac{\partial PO_4}{\partial t}}{Nitrogen Fixation} = \frac{Nfix}{A} - Denit}$$
Nitrogen Fixation
by Trichodesnium Denitrification: when O₂ reduces,
nitrate is consumed during OM
remineralization instead of O₂
Denit = $R_{NO3}\lambda_{DOC}^*(1 - \Delta(O_2))DOC$



Oxygen equation :

$$\frac{\partial O_{2}}{\partial t} = O_{2}^{ut}(\mu_{NH_{4}}^{P}P + \mu_{NH_{4}}^{D}D) + (O_{2}^{ut} + O_{2}^{nit}))$$

$$(\mu_{NO_{3}}^{P}P + \mu_{NO_{3}}^{D}D) + O_{2}^{nit}N_{fix}$$

$$-O_{2}^{ut}\gamma^{Z}(1 - e^{Z} - \sigma^{Z})\sum_{I}g^{Z}(I)Z - O_{2}^{ut}\gamma^{M}$$

$$Respiration by zooplancton and higher trophic levels (proportional to grazing)$$

$$(1 - e^{M} - \sigma^{M})\left(\sum_{I}g^{M}(I) + \sum_{I}g_{FF}^{M}(I)\right)M - O_{2}^{ut}\gamma^{M}R_{up}^{M}$$

$$-O_{2}^{ut}Remin - O_{2}^{nit}Nitrif$$

$$Consumption of oxygen by nitrification (NH_{4}->NO_{3})$$

$$Production by photosynthesis (new prod., regenerated prod., N fixation)$$

$$Respiration by zooplancton and higher trophic levels (proportional to grazing)$$

 O_2^{ut} , O_2^{nit} = different O/C Redfield ratios for new and regenerated production

At depth (out of euphotic layer) : $\partial_t O_2 \sim -O_2^{ut}$ Remin $-O_2^{Nit}$ Nitrif



Oxygen equation :

$$\frac{\partial O_2}{\partial t} = O_2^{ut}(\mu_{NH_4}^P P + \mu_{NH_4}^D D) + (O_2^{ut} + O_2^{nit})$$

$$(\mu_{NO_3}^P P + \mu_{NO_3}^D D) + O_2^{nit} N_{fix}$$

$$- O_2^{ut} \gamma^Z (1 - e^Z - \sigma^Z) \sum_I g^Z (I) Z - O_2^{ut} \gamma^M$$
Biogeochemical processes
$$(1 - e^M - \sigma^M) \left(\sum_I g^M (I) + \sum_I g^M_{FF} (I) \right) M \cdot - O_2^{ut} \text{Remin} - O_2^{nit} \text{Nitrif}$$

$$- U \cdot \partial_x O_2 - V \cdot \partial_y O_2 - W \cdot \partial_z O_2 + \partial_z (K \partial_z O_2) + \text{Diff}_h (O_2) + F_{atm}$$
Physical processes
$$3D \text{ advection} \quad \text{vertical mixing} \quad \text{horizontal air-sea mixing} \quad \text{flux}$$

...same physical terms for all of PISCES tracers



Many parameters in PISCES....

Parameter	Units	Value	Description									
				Parameter	Units	V	/alue	Description				
$\mu_{\rm max}^{\rm o}$	d^{-1}	0.6	Growth rate at 0 °C	b_Z	-	1	.079; 1.079	Temperature s	ensitivi	ity term		
μ_{ref}	d ⁻¹	1.0	Browth fate reference for light limitation	e_{\max}^{I}	-	0	3; 0.35	Maximum gro	wth eff	ficiency of zoo	plankton	
bresp b p	u _	1.066	Temperature sensitivity of growth	vI	_	0	6:0.6	Excretion as DOM				
aI	$(Wm^{-2})^{-1}d^{-1}$	2.2	Initial slope of $P = I$ curve	gm	d^{-1}	3	;0.75	Maximum gra	zing ra	te		
δ^{I}	(wm) u	0.05:0.05	Exudation of DOC	g_{FF}^M	(mmolL ⁻¹) ⁻¹ 2	$\times 10^3$	Flux feeding	ate			
BI	_	2.1:1.6	Absorption in the blue part of light	$K_{\mathcal{G}}^{I}$	μ molCL ⁻¹	2	0;20	Half-saturatio	n const	ant for grazing	3	
BI	_	0.42:0.69	Absorption in the green part of light	P'P	-	1	;0.3	Preference for	nanop	hytoplankton		
B1	_	0.4:0.7	Absorption in the red part of light	^P D n	_	0	103	Preference for	POC	115		
$K^{I,\min}$	nmol PL -1	08.24	Minimum half-saturation constant for phosphate	p_{Z}^{PPOC}	_	1	.0	Preference for	microz	zooplankton		
K ^I ,min	$umol NL^{-1}$	0.013:0.039	Minimum half-saturation constant for phosphate	Fthresh	µmolCL ⁻¹	0	.3; 0.3	Food threshol	d for zo	ooplankton		
KNO	μ mol N L ⁻¹	0.13; 0.39	Minimum half-saturation constant for nitrate	J_{thres}^{L} J_{thres}^{M}	µmolCL ⁻	0	.001	Specific food	thresho	lds for microz	ooplankton	
$K_{si}^{D,min}$	µmol SiL ⁻¹	1	Minimum half-saturation constant for silicate	m	(µmol CL	Paramete	er Units		N	Value	Description	
KSi	µmol SiL ⁻¹	16.6	Parameter for the half-saturation constant	r' K	d ⁻¹	λ_{POC}	d^{-1}		0).025	Degradation rate of POC	
$K_{S_i}^I$	µmol SiL ⁻¹	2;20	Parameters for Si / C	v^{I}		w _{POC}	$m d^{-1}$		2	2	Sinking speed of POC	
$K_{F_{2}}^{I,\min}$	nmol Fe L ⁻¹	1:3	Minimum half-saturation constant for iron uptake	$\theta^{\text{Fe},\text{Zoo}}$	µmolFe n	wmin	$m d^{-1}$		3	30	Minimum sinking speed of GOCh	
$S_{rat}^{f^{e}}$	_	3;3	Size ratio of Phytoplankton			wdust	$m s^{-1}$		2	2	Sinking speed of dust	
$\theta_{m}^{Si,D}$	mol Si (mol C) ⁻¹	0.159	Optimal Si / C uptake ratio of diatoms			a ₆	(µmol	$CL^{-1})^{-1}d^{-1}$	-1 2	25.9	Aggregation rate (turbulence) of PC	C→GOC
here, I	μ mol Fe (mol C) ⁻¹	7:7	Optimal iron quota			a7	(µmol	$CL^{-1})^{-1}d^{-1}$	-1 4	1452	Aggregation rate (turbulence) of PC	C→GOC
$\theta^{\text{Fe},I}$	μ mol Fe (mol C) ⁻¹	40.40	Maximum iron quota			as	(µmol	$CL^{-1})^{-1}d^{-1}$	-1 3	3.3	Aggregation rate (settling) of POC-	→GOC
max mI	d ⁻¹	0.01.0.01	phytoplankton mortality rate			ag	(µmol	$CL^{-1})^{-1}d^{-1}$	-1 4	47.1	Aggregation rate (settling) of POC-	→GOC
m ^P	$d^{-1} \mod C^{-1}$	0.01	Minimum quadratic mortality of phytoplankton			$\lambda_{E_2}^{min}$	d^{-1}		3	3×10^{-5}	Minimum scavenging rate of iron	
w^D	$d^{-1} \operatorname{mol} C^{-1}$	0.03	Maximum quadratic mortality of diatoms			λ _{Fe}	$d^{-1}\mu$	nol ⁻¹ L	0	0.005	Slope of the scavenging rate of iron	xlam1
$\theta_{\text{Chl},I}^{\text{max}}$	$m_{g}Chl(m_{g}C)^{-1}$	0.033.0.05	Maximum Chl / C ratios of phytoplankton			λdust	$d^{-1}m$	$g^{-1}L$	1	150	Scavenging rate of iron by dust	
θ Chl	$mgChl(mgC)^{-1}$	0.0033	Minimum Chl / C ratios of phytoplankton			λCaCOa	d^{-1}	0	0).197	Dissolution rate of calcite	
^o min Imax	μ mol CL ⁻¹	1:1	Threshold concentration for size dependency			nca	-		1	l	Exponent in the dissolution rate of o	calcite
- max	µ	-,-	The should concentration for one acpendency			Xlab	_		0).5	Proportion of the most labile phase	in PSi
						λ slow	d^{-1}		0).003	Slow dissolution rate of BSi	
						λfast	d^{-1}		0	0.025	Fast dissolution rate of BSi	
						131						
						Parameter	Units	V	alue	Descripti	on	
Parameter	Units	Value	Description			λ_{NH_4}	d^{-1}	0.	05	Maximur	n nitrification rate	
			1			O2min,1	µmol O ₂ I	2-1 1		Half-satu	ration constant for denitrification	
λ_{DOC}	d ⁻¹	0.3	Remineralization rate of DOC			O2 ^{min,2}	µmol O ₂ I	2-1 6		Half-satu	ration constant for denitrification	
KDOC	μ mol CL ⁻¹	417	Half-saturation constant for DOC remin.			L _T M ^m	nmol NI	-1d-1 = 0	012	Total con Maximur	centration of iron ligands	
KBact	μ mol N L ⁻¹	0.03	NO3 half-saturation constant for DOC remin			KDz	nmol Fe I	-1 0.	1	Fe half-s	aturation constant of nitrogen fixation	
KBact	μ mol N L ⁻¹	0.003	NH4 half-saturation constant for DOC remin			E _{fix}	Wm^{-2}	. 50)	Photosyn	thetic parameter of nitrogen fixation	
KBact	$\mu mol PL^{-1}$	0.003	PO4 half-saturation constant for DOC remin			Feice	nmol Fe L	-1 1:	5	iron conc	entration in sea ice	
K _{Ea}	nmol Fe L ⁻¹	0.01	Fe half-saturation constant for DOC remin.			Fe,min SolEs	µmol Fe n –	1 ~ d ~ 1 2 0.	02	Maximur Solubility	n sediment flux of Fe	
a1	$(\mu mol CL^{-1})^{-1}$	d^{-1} 0.37	Aggregation rate (turbulence) of $DOC \rightarrow POC$	2		O ₂ ^{ut}	mol O ₂ (n	nol C) ⁻¹ 13	3/122	O/C for	ammonium-based processes	
<i>a</i> 2	$(umol CL^{-1})^{-1}$	d^{-1} 102	Aggregation rate (turbulence) of $DOC \rightarrow POC$	~		O2nit	mol O ₂ (n	$(100 {\rm C})^{-1}$ 32	2/122	O / C rati	o of nitrification	
	(umol CI =1)=1	d=1 2520	Aggregation rate (turbulence) of DOC > CO	c		$r_{NH_4}^{\star}$	mol N (m	olC) ⁻¹ 3/	5	C/N ratio	o of ammonification	
uz	(uniorCL ·) ·	u 5550	Aggregation rate (turbulence) of $DOC \rightarrow GO$	C		r*	mol N (m	olC) ⁻¹ 10)5/16	C/N ratio	of denitrification	

	Aum	ont et a	I., 20)15
Aggregation rate (Brownian) of DOC \rightarrow POC	r _{CaCO3}	-	0.3	Rain-ra
Aggregation rate (Brownian) of $DOC \rightarrow POC$	$r_{NO_3}^{\star}$ $_{ heta^{N,C}}$	$mol N (mol C)^{-1}$ $mol N (mol C)^{-1}$	105/16 16/122	C/N ra
Aggregation rate (furbulence) of $DOC \rightarrow GOC$	14114			

16/122 N / C Redfield ratio

Rain-ratio parameter

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 $(\mu mol CL^{-1})^{-1}d^{-1}$

 $(\mu mol CL^{-1})^{-1}d^{-1}$

 a_4

 a_5

5095

114

More sophisticated PISCESv2 options : PISCES-quota

PISCES-QUOTA (39/40 tracers) 24/25 in PISCES std



More sophisticated PISCESv2 options : Ligands for Fe

- Dissolved Fe assimilated by phyto mostly in its complexed for with ligands (L)

- ligand concentration is constant in previous version v0,v1

=> prognostic equation for1 ligand concentration in v2 if chosen (other version with more L)

More sophisticated PISCESv2 options : sediment module

 In the default configuration, exchanges with the sediments are modeled based on a simple metamodel proposed by Middelburg et al. (1996):

$$F_{sed} = F(NO_3, O_2, Z, ...)$$

New process model : new chemical species : Sulfate, FeS,... dissolved and precipitate

=> long integration time (~100s of years) to reach equilibirum
 => very new in CROCO (not yet used), soon in West Africa (Senegal) (P.-A. Auger, IRD/LOPS)

More sophisticated PISCESv2 options : diurnal vertical migration of zooplankton (Gorgues et al., 2019)

- Not a prognostic parameterization !
- DVM parameterization is activated by ln_dvm_meso = .true.
- Migration depth is parameterized according to Bianchi et al. (2013)

 $Z_{mig} = F(O2, Chl, T)$

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More sophisticated PISCESv2 options : diurnal vertical migration of zooplankton

- A constant fraction of mesozoo is prescribed to migrate (xfracmig). Microzoo is not migrating
- Organisms are assumed to be at the surface at night and at the migration depth during daytime
- Organisms are supposed to respire, excrete DOM and inorganic nutrients and egest fecal pellets in both habitats (function of daylength and temperature)

Model structure and routine names

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End of part 2

